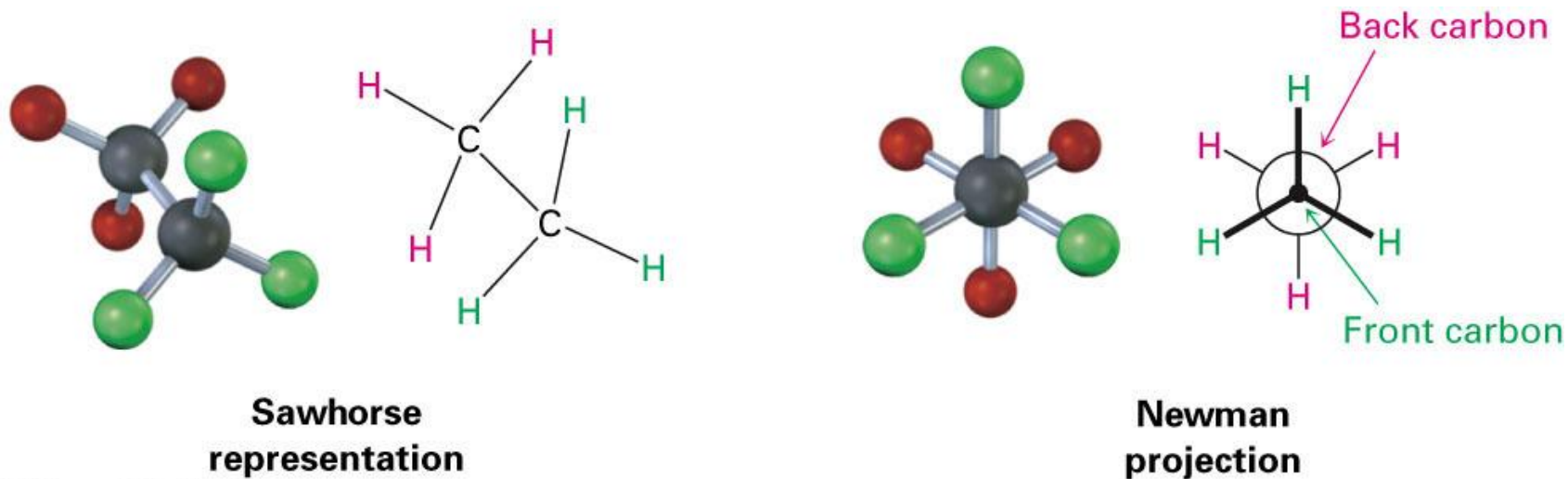


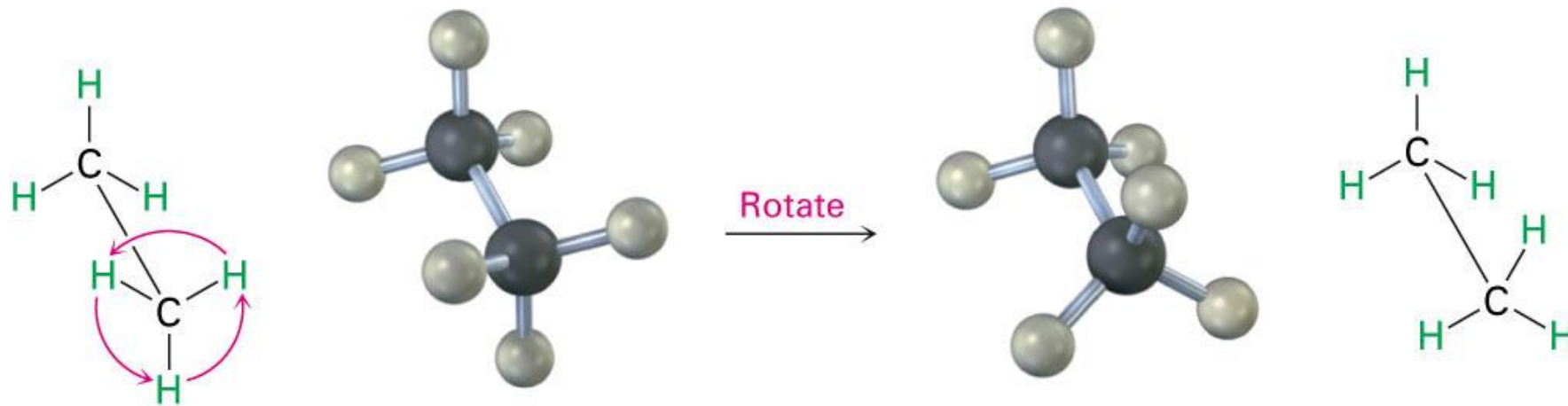
- **Conformation**- Different arrangement of atoms resulting from bond rotation
- Conformations can be represented in 2 ways:



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- We do not observe perfectly free rotation
- There is a barrier to rotation, and some conformers are more stable than others
- **Staggered**- most stable: all 6 C-H bonds are as far away as possible
- **Eclipsed**- least stable: all 6 C-H bonds are as close as possible to each other

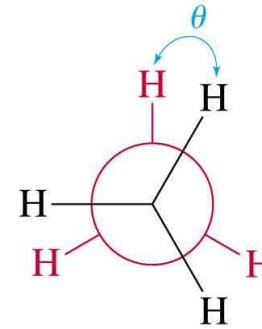
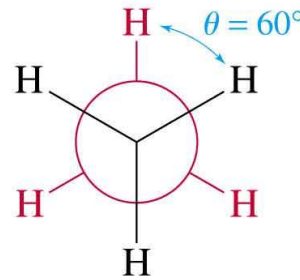
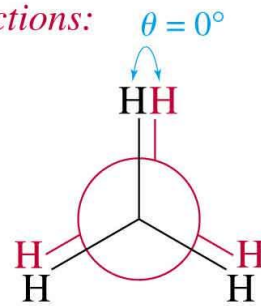
- Stereochemistry concerned with the 3-D aspects of molecules
- Rotation is possible around C-C bonds in open-chain molecules (not cyclic)



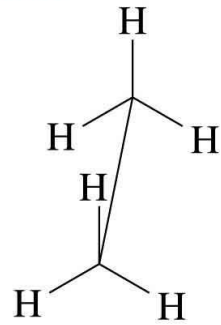
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- Eclipsed conformer has highest energy
- Dihedral angle = 0 degrees

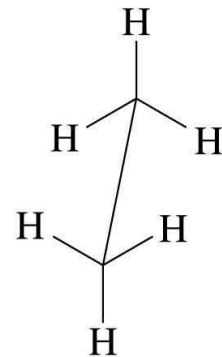
Newman projections:



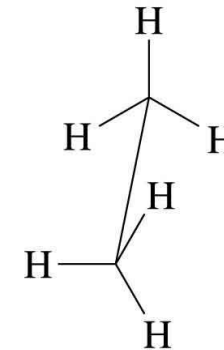
Sawhorse structures:



eclipsed, $\theta = 0^\circ$

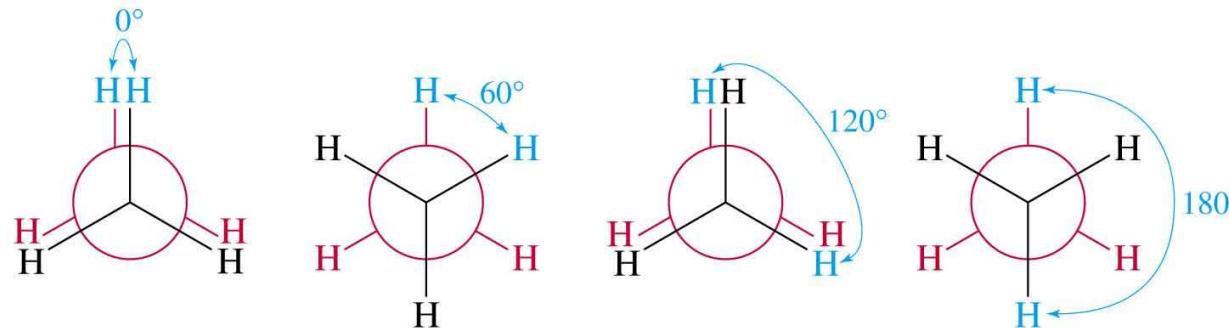
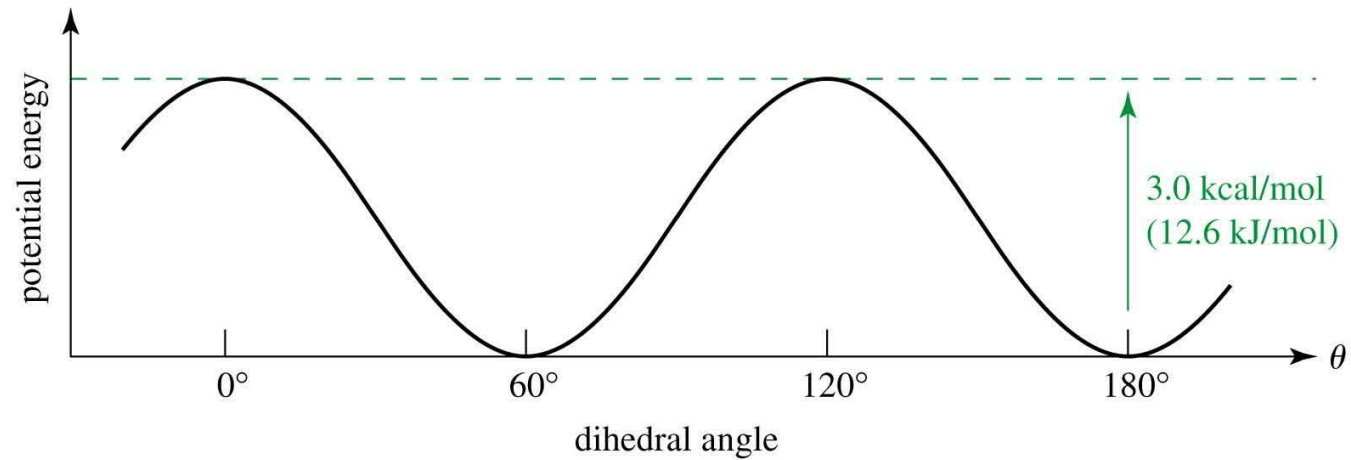


staggered, $\theta = 60^\circ$

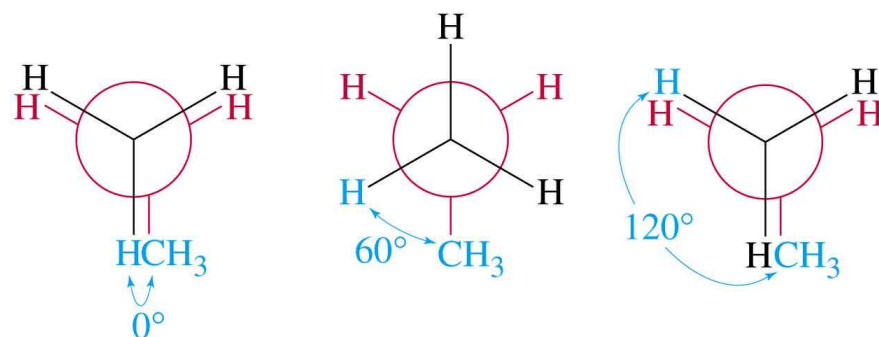
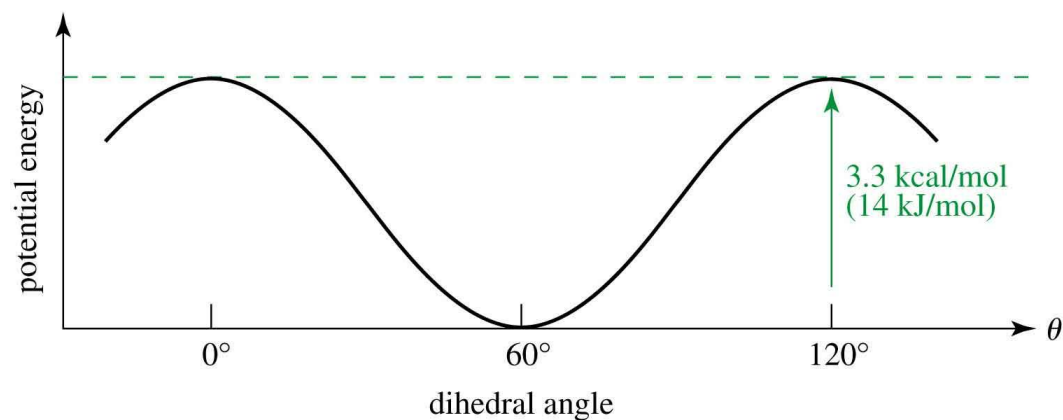


skew, $\theta = \text{anything else}$

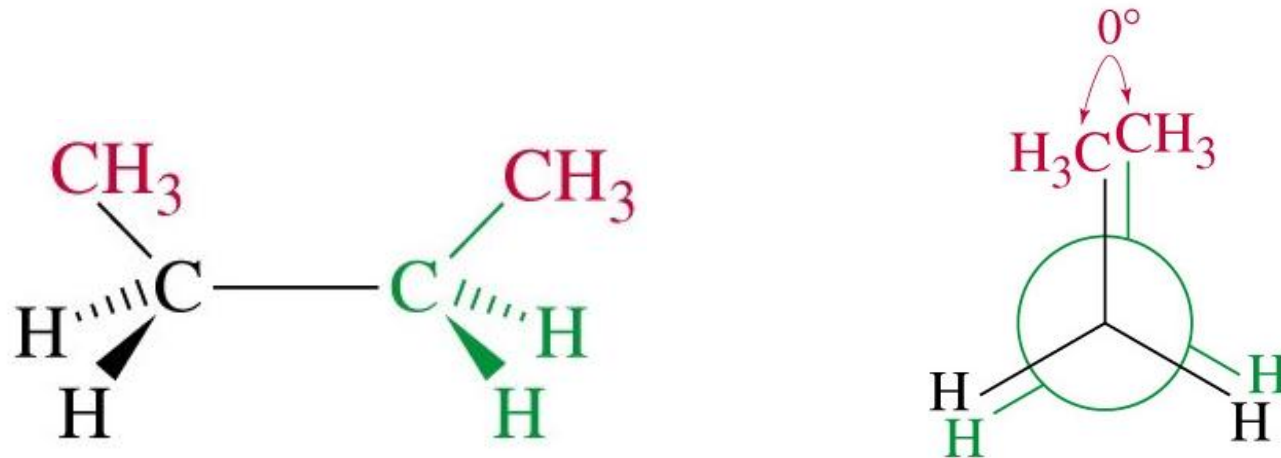
- Torsional strain: resistance to rotation.
- For ethane, only 3.0 kcal/mol



Note slight increase in torsional strain due to the more bulky methyl group.

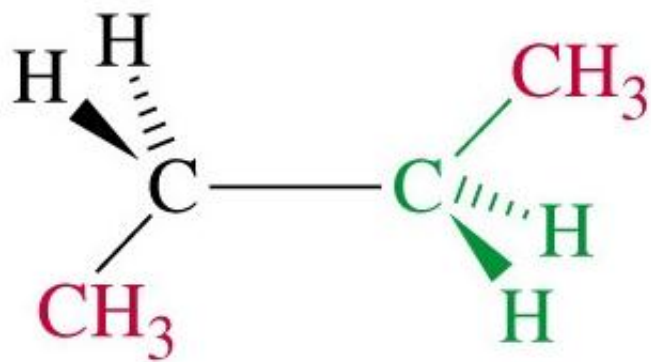


- Highest energy has methyl groups eclipsed.
- Steric hindrance
- Dihedral angle = 0 degrees

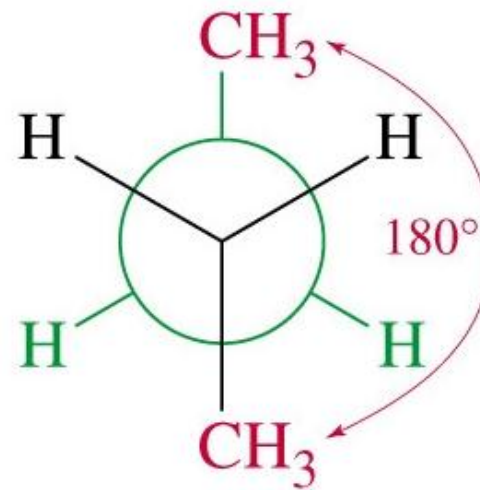


totally eclipsed

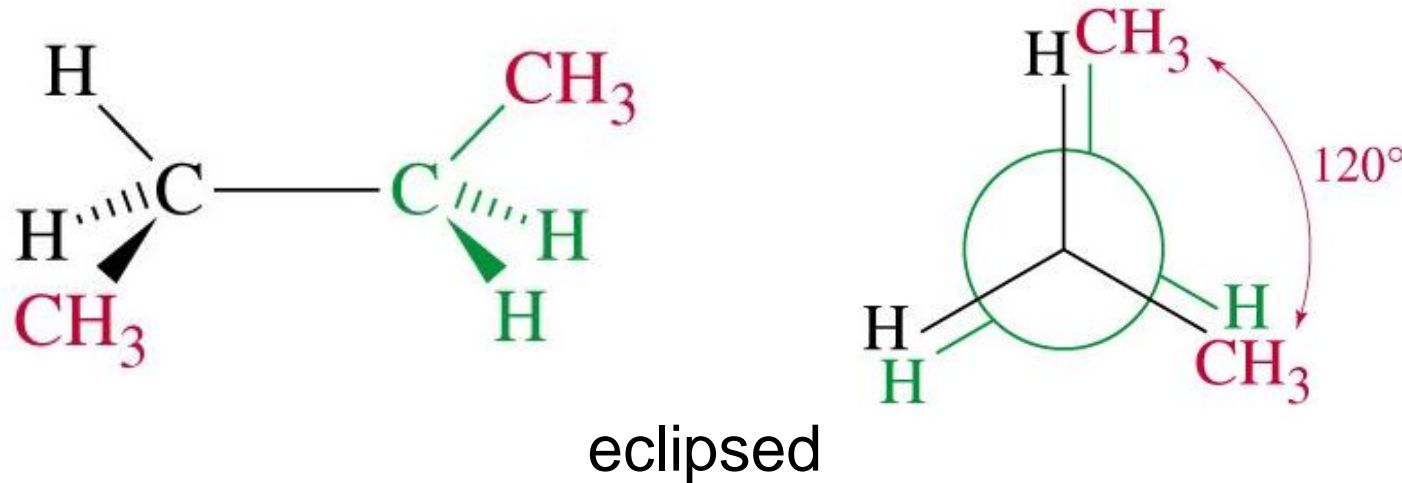
- Lowest energy has methyl groups **anti**.
- Dihedral angle = 180 degrees



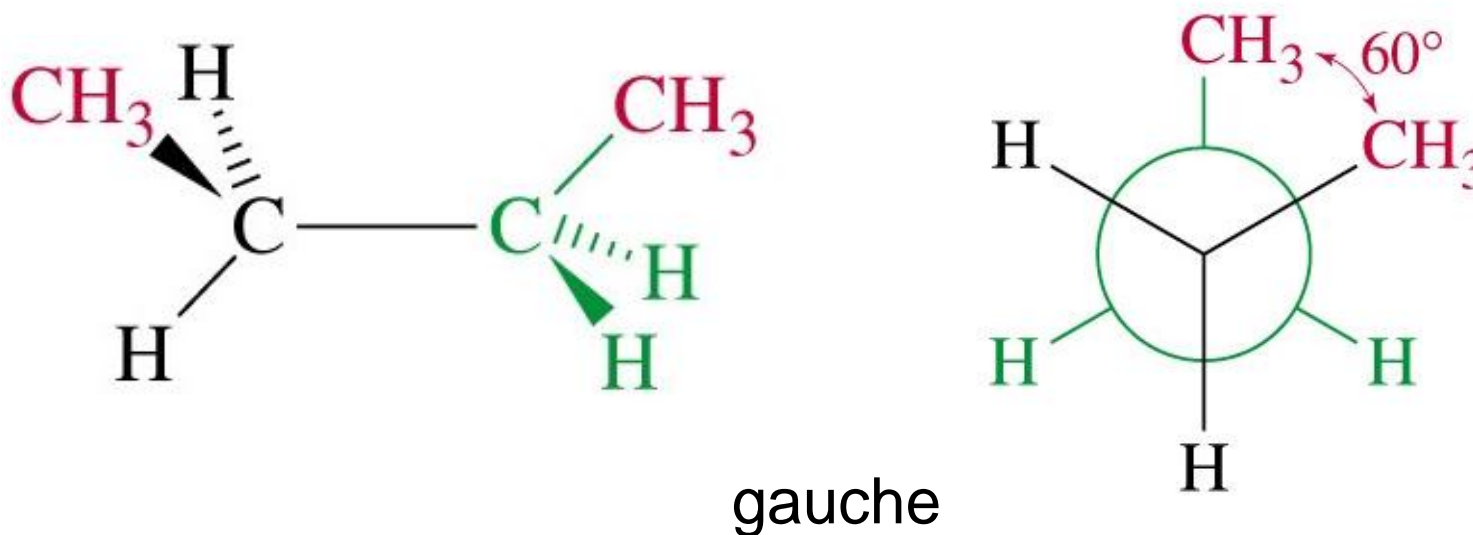
anti



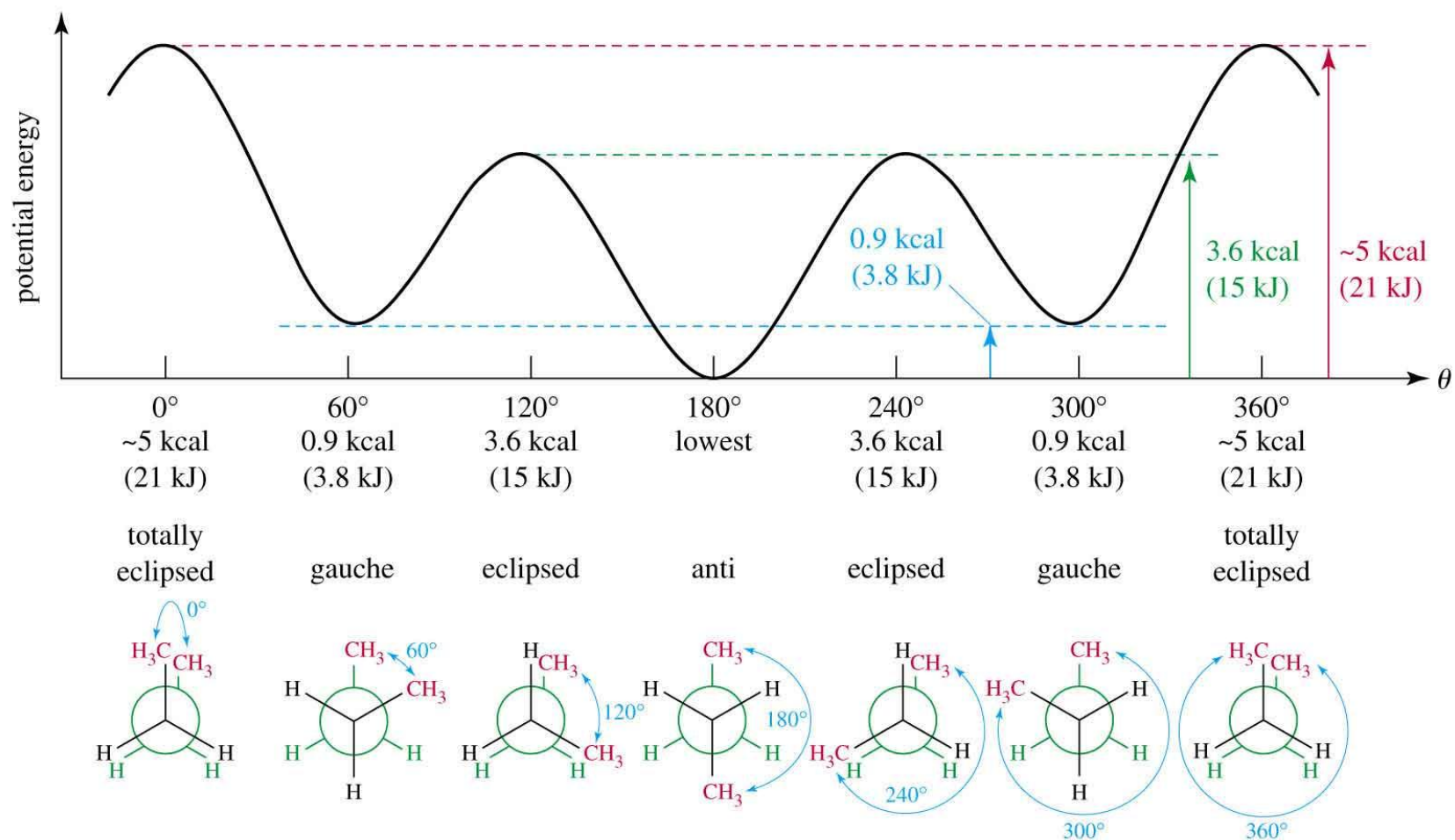
- Methyl groups eclipsed with hydrogens
- Higher energy than staggered conformer
- Dihedral angle = 120 degrees



- **Gauche**, staggered conformer
- Methyls closer than in anti conformer
- Dihedral angle = 60 degrees



Conformational Analysis



- Anti conformation is lowest in energy.
- “Straight chain” actually is zigzag.

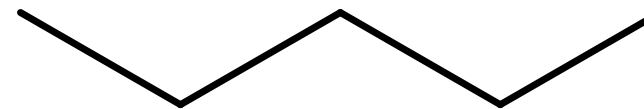
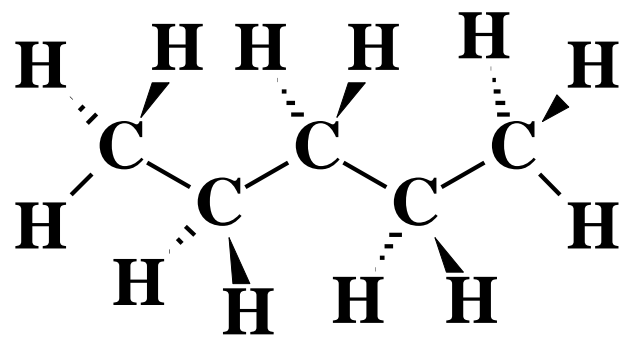


Table 3.5 | **Energy Costs for Interactions in Alkane Conformers**

Interaction	Cause	Energy cost	
		(kJ/mol)	(kcal/mol)
H ↔ H eclipsed	Torsional strain	4.0	1.0
H ↔ CH ₃ eclipsed	Mostly torsional strain	6.0	1.4
CH ₃ ↔ CH ₃ eclipsed	Torsional and steric strain	11	2.6
CH ₃ ↔ CH ₃ gauche	Steric strain	3.8	0.9

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