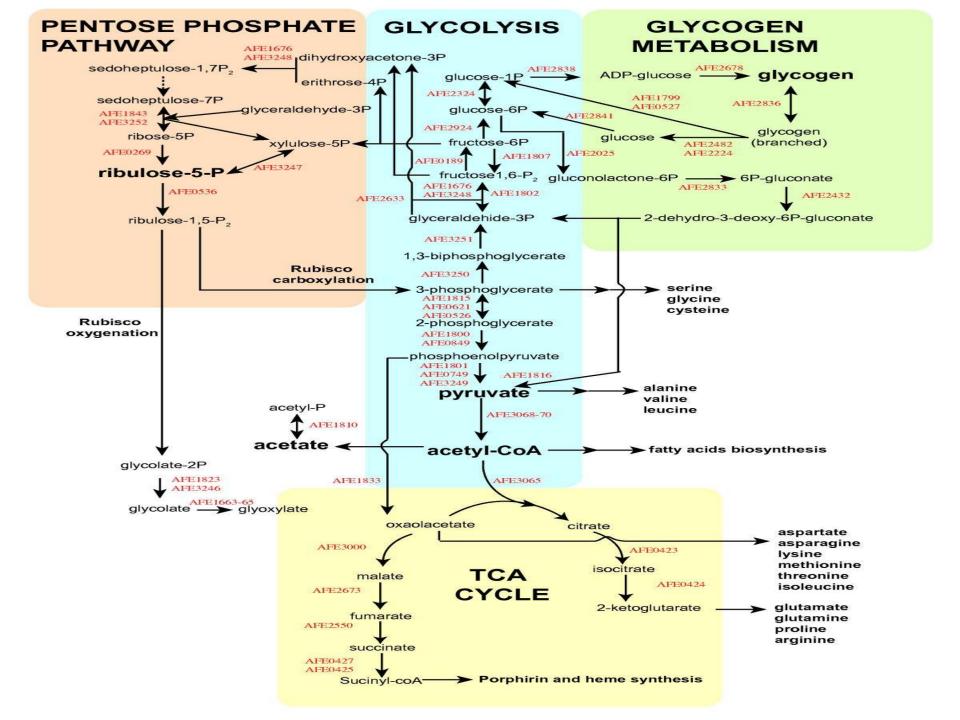
Lecture 1 Session Three Carbohydrate Metabolism 2 Dr. Khawla A. Shemran



Pentose Phosphate Pathway (PPP) or Hexose Monophosphate Shunt (HMS)

It is characterized by:

- carried out in the cytosol
- It is an alternative pathway for glycolysis that does not produce energy.
- Its enzymes (like glycolysis) are in the cytosol but the hydrogen acceptor is NADP⁺ instead of NAD⁺.

functions

- It has three main functions:
- **1- Generates NADPH (reducing equivalent) for:**
- biosynthetic/anabolic pathways such as fatty acid and steroid hormone synthesis.
- Protection against oxidants (see later)
- 3- **Re- production of glucose** and other Hexoses from pentose sugar.

Phases (2 phases) oxidative & Rearrangement

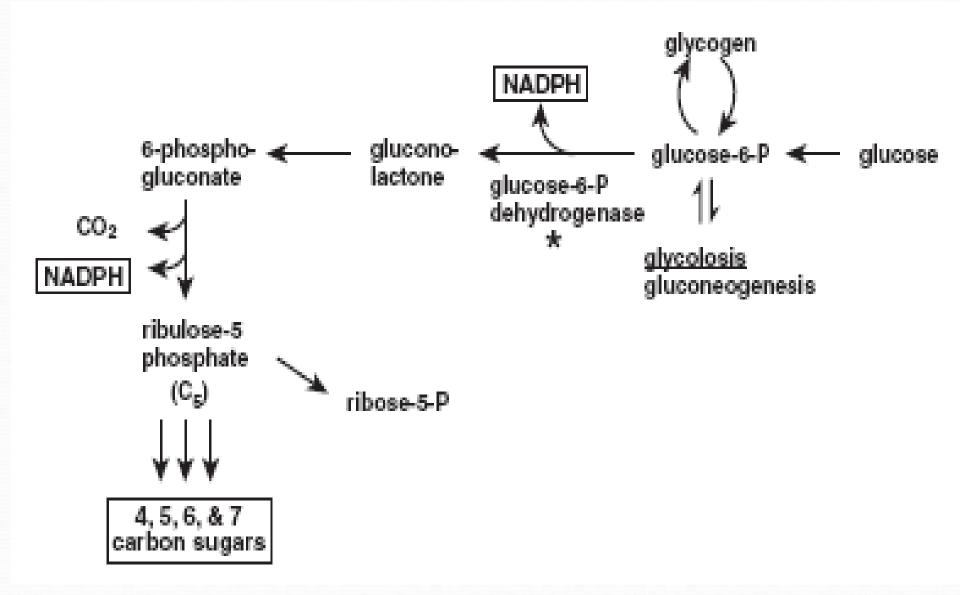
In phase one:

Glucose-6-phosphate + 2 NADP+ $\rightarrow \rightarrow \rightarrow$ C5-sugar phosphate + 2 NADPH + 2H+ + CO2

• In phase two:

3C5-sugar phosphate $\rightarrow \rightarrow \rightarrow \rightarrow \rightarrow 2$ fructose-6phosphate and glyceraldehyde 3-phosphate $\rightarrow \rightarrow \rightarrow \rightarrow \rightarrow \rightarrow$ Glucose again

Two Phases



Check point

What is the source of NADPH? Which organs in human rich in this metabolic pathway?

Answers

- The major source of NADPH is the <u>pentose phosphate pathway</u>.
 Lesser-known mechanisms are:
- NADP-linked malic enzyme
- NADP-linked <u>isocitrate dehydrogenase</u>
- This pathway is active in:
- liver,
- adipose tissue,
- adrenal cortex,
- testes and
- lactating mammary glands.

Role of NADPH in RBCs

- In RBCs, reactive oxygen species (superoxide and hydrogen peroxide) are formed normally during the process of oxygen transport as in the following:
 - Hb-Fe+2 -O2 \rightarrow Hb-Fe+3 + O2.

(superoxide anion)

 This reaction is spontaneous and it takes place 1% per hour Superoxide anion is very reactive and harmful to the cell due to the formation of free radicals, it inactivated by:

$\mathbf{2O2} \cdot \mathbf{\overline{+2H+}} \rightarrow \mathbf{H2O2+O2}$

- Hydrogen peroxide can also form free radicals, so that it must be inactivated by glutathione (GSH)
- $2GSH + H_2O_2 \rightarrow G-S-S-G + 2H_2O$
- Regeneration of GSH must be carried out to maintain sufficient level for normal metabolism by:

 $G-S-S-G + 2NADPH \rightarrow 2GSH + 2NADP+$

• **Sufficient level of NADHP** is crucial for normal metabolism of RBC.

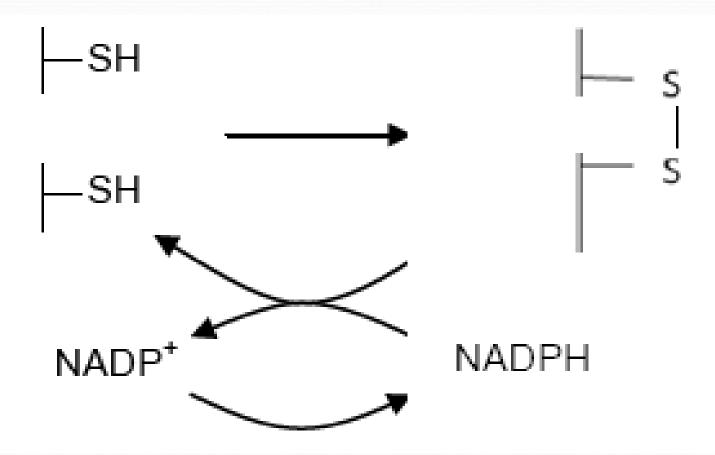
Which control G6PD synthesis

- The gene encodes G6PD is found on the long arm of the X chromosome .
- In some individuals this gene is mutated
- So it is <u>X-linked recessive</u> <u>hereditary disease</u>

What happens when G6PD gene is mutated?

- G6PD is low
- H2O2 accumulates in RBCs
- Leads to oxidation of hemoglobin and other proteins.
- Hemoglobin becomes cross-linked by disulphide bonds to form insoluble aggregates called Heinz bodies as in the following

Oxidation of hemoglobin



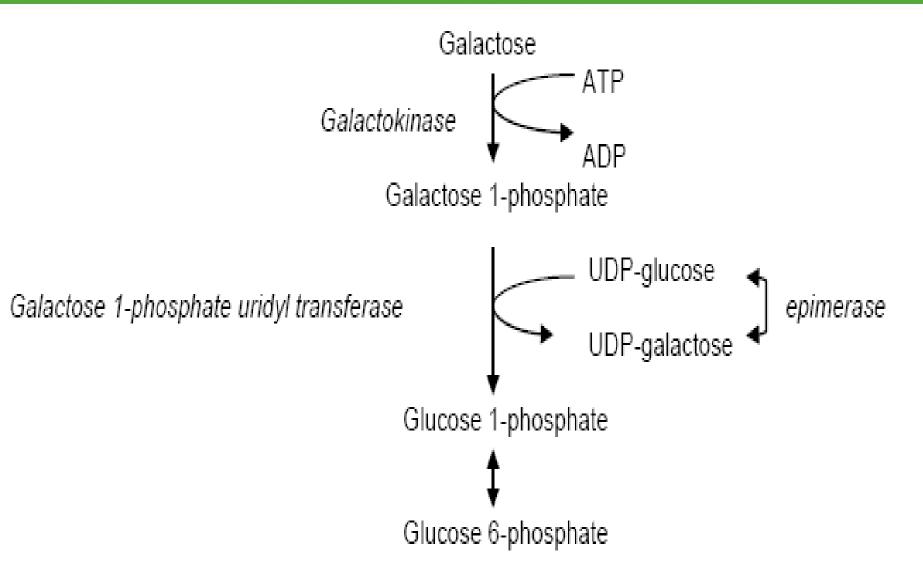
What happens when G6PD gene is mutated?

- Oxidation of Hb leads to premature destruction of RBCs (fragile) and causes haemolysis (Acute haemolytic anemia)
- It is **precipitated** by chemicals that increase the oxidative stress (e.g. sulphonamides, aspirin, NSAIDs and antimalarials),
- It is also **precipitated** when an individual have eaten fava beans which contain glycosides .
- For this reason it is called **favism**

Consequences of G6PD deficiency

- G6PD gene mutation
- No or Low NADPH
- GSSG not return back to GSH
- RBC membrane rigid and fragile
- Mechanical destruction of RBCs leads
- Low Hb----- anemia
- High bilirubin----- Jaundice

Galactose Metabolism

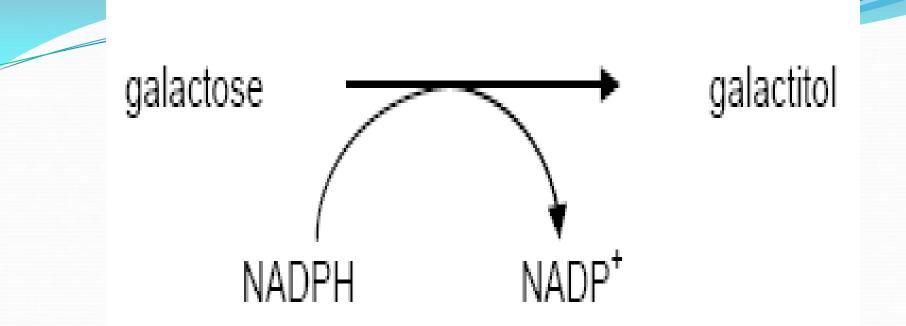


Galactosemia

- is a rare genetic metabolic disorder that affects an individual's ability to metabolize the diet sugar galactose
- <u>Lactose</u> in food (such as dairy products) is broken down by the enzyme <u>lactase</u> into <u>Glucose</u> and <u>Galactose</u>.
- In individuals with galactosemia, the enzymes needed for further metabolism of galactose (*kinase* or *transferase*) are severely diminished or missing entirely,
- Leading to accumulation of toxic levels galactose and /or Gal-1-P

Galactosemia

- The absence of the <u>kinase</u> is relatively rare and is characterized by accumulation of <u>galactose only</u> in tissues.
- The absence of the **transferase** is more common and more serious as **both** galactose and galactose 1-phosphate accumulate in tissues.
- Accumulation of G1P in tissues leads to its reduction to <u>galactitol</u> (aldehyde group reduced to alcohol group) by the activity of the enzyme *aldose reductase*:



 Galactitol is not a substrate for next enzyme the in the polyol pathway, <u>polyol dehydrogenase</u> ---increase osmolarity---- cells rupture (damage)

 This reaction depletes some tissues of NADPH.---no reducing equivalent---protein cross linking (damage)

Clinical Effects

• Eye Damage:

- Cataract: due to-- cross linking of lens proteins by -S-Sbond formation and nonenzymatic glycosylation of the lens proteins because of the high concentration of galactose
- 2. Glucoma; Accumulation of galactose and galactitol in the eye may lead to raised intra-ocular pressure (glaucoma) which if untreated may cause blindness

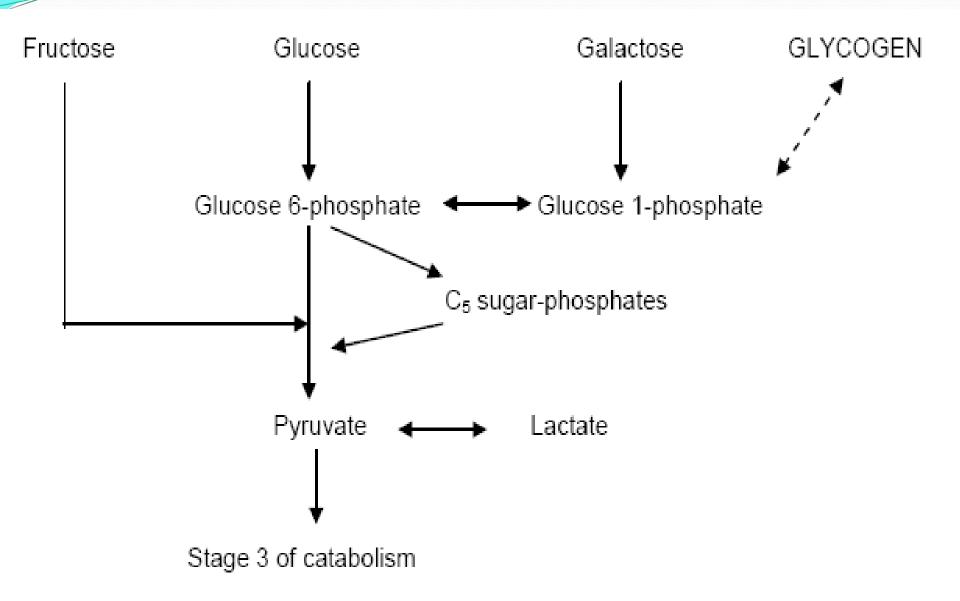
• Other tissues:

- Accumulation of galactose 1-phosphate resulting in <u>hepatomegaly</u> (an enlarged <u>liver</u>), <u>cirrhosis</u>, <u>renal failure</u>, <u>cataract</u>, <u>brain damage</u>, <u>and ovarian failure</u>.
- This related to the **sequestration** of Pi making it unavailable for ATP synthesis.

Metabolism of Pyruvate

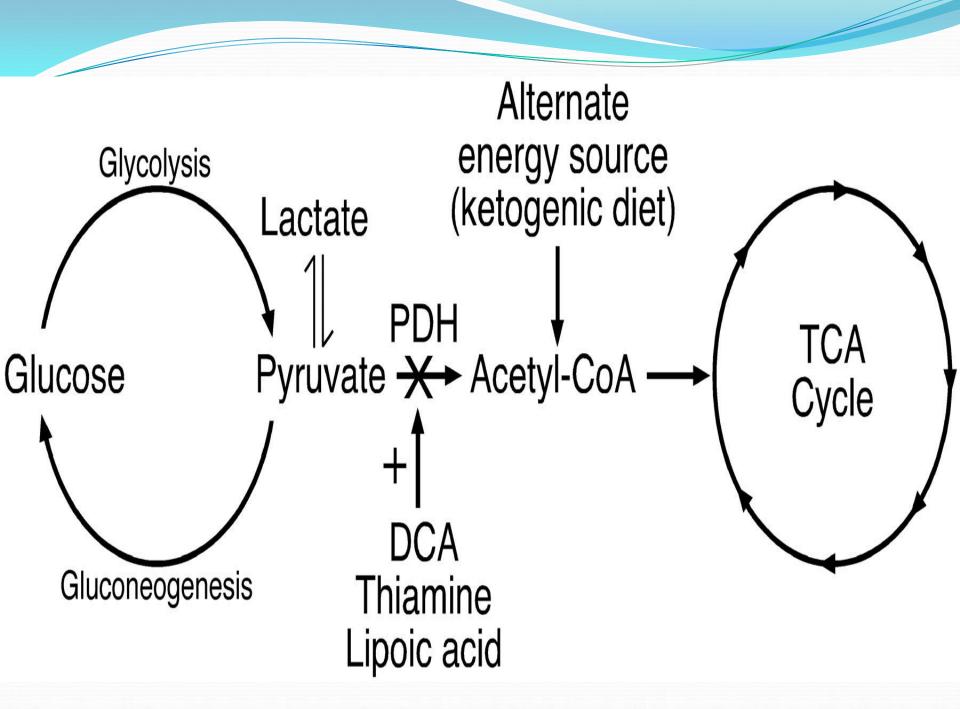
- To enter stage 3 of catabolism, Pyruvate is first converted to acetyl~CoA by the enzyme *pyruvate dehydrogenase* (*PDH*).
- PDH is a multi-enzyme complex that catalyses the overall reaction:
- Pyruvate + CoA + NAD+ \rightarrow Acetyl~CoA + CO₂ + NADH + H+
- This reaction is irreversible
- Is PDH sensitive to vitamin B deficiency?

Summary of stage 2 catabolism of sugars



Control mechanisms of PDH reaction

- The acetyl~CoA inhibits PDH so only Acetyl~CoA from the β-oxidation of fatty acids rather than from glucose is used in stage 3 of catabolism.
- The reaction is sensitive to the energy status of the cell (ATP and NADH inhibit and ADP activates the enzyme)
- The enzyme is activated when there is plenty of glucose to be catabolised (insulin activates the enzyme by promoting its dephosphorylation).



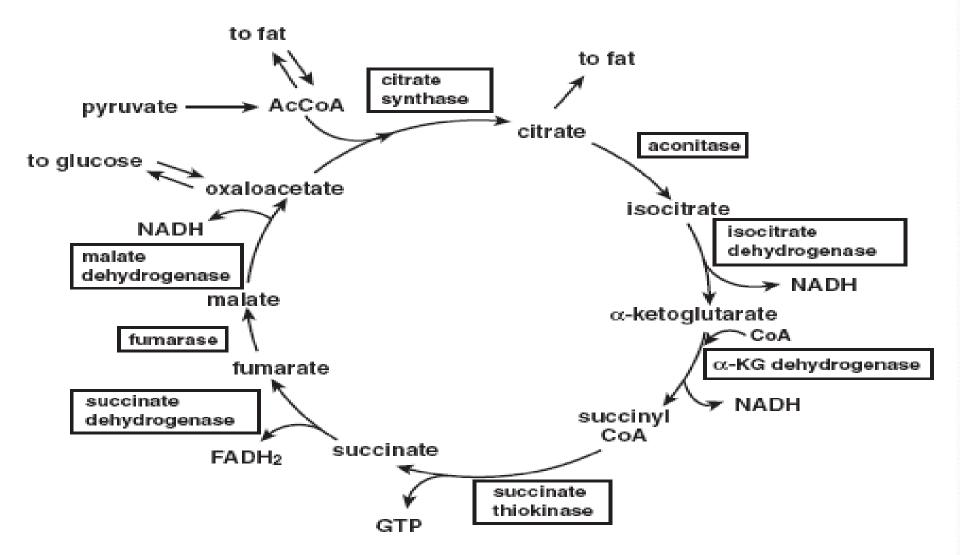
Lecture 2 Session 3 Tricarboxylic Acid Cycle & Gluconeogenesis

Dr. Khawla A. Shemran



Tricarboxylic Acid Cycle (TCA)

This is the stage 3 of CHO metabolism. It is an oxidative pathway that occurs in **mitochondria** The overall equation for the pathway is:



Functions of TCA

- <u>Catabolism</u>: of sugars, fatty acids, ketone bodies, alcohol and amino acids leads to *formation* of acetyle CoA and CO₂. (Main function)
- The H+ and e- removed from **acetate** are transferred to NAD+ and FAD (stage 4 **energy production**).

Anabolic (**biosynthetic**) of the intermediates include:

- α-ketoglutarate, succinate, fumarate and malate used for the synthesis of nonessential amino acids.
- Succinate and oxaloacetate used in synthesis of heme and glucose
- Citrate used in synthesis of **fatty acids**.

Regulation of the TCA cycle

- Two major signals feed information on the rate of utilization of ATP to the TCA cycle:
 - ATP/ADP ratio
 - NADH/NAD+ ratio
- One of the early irreversible steps of the TCA cycle (catalysed by *isocitrate dehydrogenase*) is
- inhibited by the high-energy signal NADH and
- activated by the low-energy signal ADP.

Gluconeogenesis (GNG)

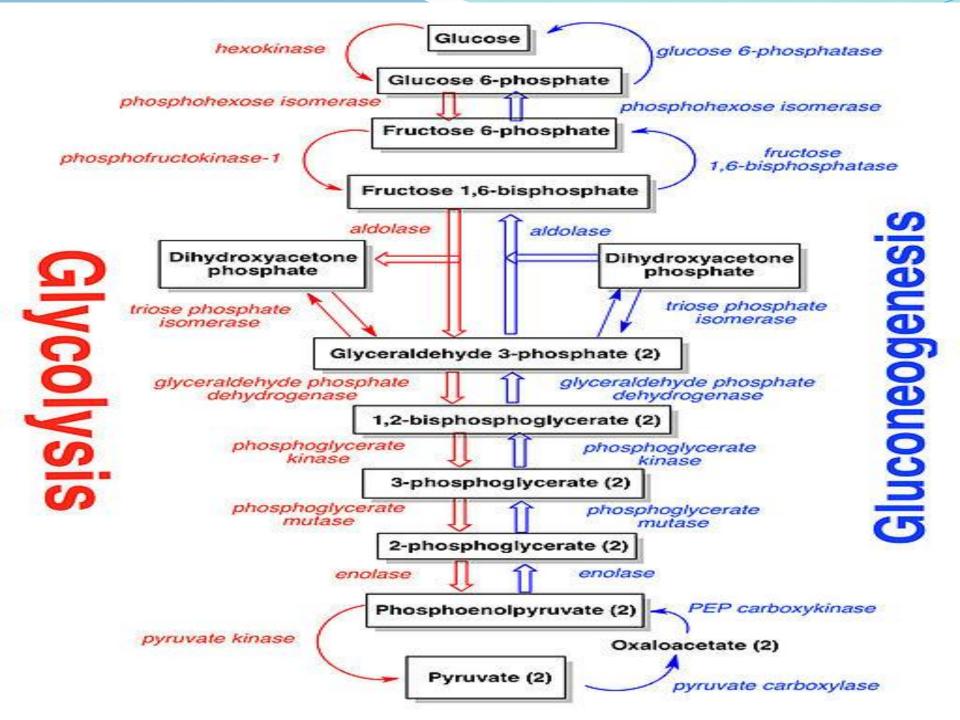
- Is a metabolic pathway that results in the generation of <u>glucose</u> from non- carbohydrate carbon substrates
- <u>such as:</u>
- pyruvate,
- lactate,
- Glycerol,
- glucogenic amino acids and
- Odd chain fatty acids (propionate)

Gluconeogenesis (GNG)

- It is one of the two main mechanisms humans use to keep blood <u>glucose</u> levels from dropping too low (hypoglycemia).
- The other means of maintaining blood <u>glucose</u> levels is through the degradation of <u>glycogen</u> (<u>glycogenolysis</u>).

Gluconeogenesis (GNG)

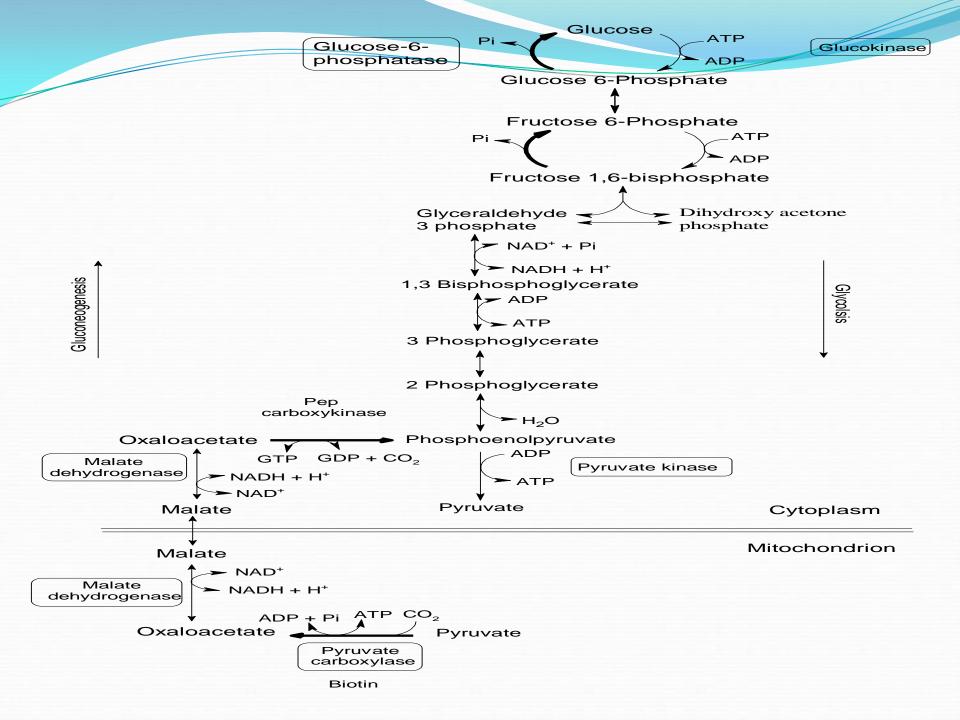
- In vertebrates, gluconeogenesis takes place mainly in the <u>liver</u> and, to a lesser extent, in the cortex of <u>kidneys</u>.
- The process occurs during periods of <u>fasting</u>, <u>starvation</u>, <u>low – carbohydrate diet or severe</u> <u>exercise</u>.
- Gluconeogenesis is also a target of therapy for type II diabetes, such as metformin, which inhibits glucose formation and stimulates glucose uptake by cells.



Why Gluconeogenesis is important?

- 1. Glucose is necessary as a **source of energy**, especially for the brain and RBCs.
- Below the critical glucose level (<45mg/dL) there is brain dysfunction which can lead to coma and death.

- Even under conditions where fat may be supplying most of the calorie requirements, there is, always basal requirements for glucose for example:
- glucose is the **only** fuel that supplies energy to the skeletal muscles under anaerobic conditions.
- Also, glucose is the precursor of **milk sugar** (lactose) in lactating mammary glands.
- **3. Gluconeogenesis** is used to clear the products of metabolism of other tissues from the blood e.g.
- lactate (produced by muscles and RBCs) and
- **glycerol** (produced by adipose tissue in glycolysis).



Regulation of gluconeogenesis

- Gluconeogenesis occurs as part of the response to stress situations (e.g. <u>fasting</u>, <u>starvation</u>, <u>prolonged exercise</u>) and is largely under hormonal control.
- The major control sites are *PEPCK* and *Fructose 1,6-bisphosphatase*. The activity of *PEPCK* is increased by *glucagon and cortisol* and decreased by *insulin*
- The activity of *Fructose 1,6- bisphosphatase* is also increased by glucagon and decreased by insulin
- The *insulin/glucagon* ratio plays a major role in determining the rate of gluconeogenesis.
- In the absence of adequate levels of biologically effective insulin, such as occurs in **diabetes**, increased rates of gluconeogenesis contribute significantly to the hyperglycaemia.