Methods of Estimating Evaporation:

- 1. Mass Transfer Methods (Myers Formula)
- 2. Energy Budget Methods
- 3. Combined Methods (Penman Equation)
- 4. Water Budget Method (Pan Evaporation Method)

1. Mass transfer method

Evaporation driven by

-Vapor pressure gradient

-Wind speed

$$E = f(u) (e_s - e_a)$$

= (a + bu) (e_s - e_a)

es: saturation vapor pressure at temperature T of the water surface

e_a: vapor pressure at some fixed level above the water surface

u : wind speed at some level above surface

a,b : empirical constants

Some formulas use a zero value for the constant "a" in the formula due to the small local air movements with velocities insufficient to remove excess vapor from a above a pan surface. Harbeck and Meyers (1970) present the following equation.

$$\mathbf{E} = \mathbf{b}\mathbf{u}_2 * (\mathbf{e}_s - \mathbf{e}_2)$$

Where:

E=Evaporation (cm/day)

b=0.012 for Lake Hefner, 0.018 for Lake Mead

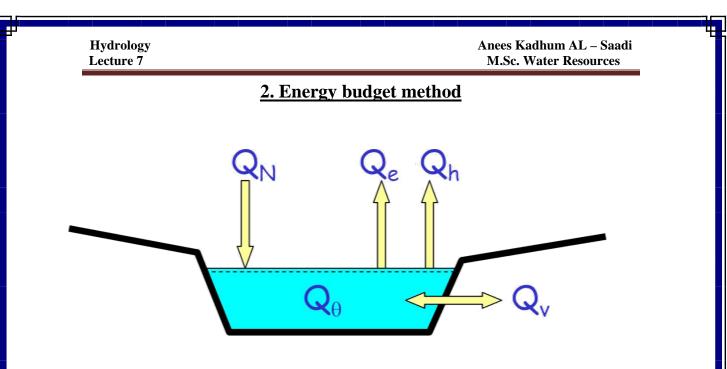
es=vapor pressure at water surface (mb)

e₂=vapor pressure 2 m above water surface (mb)

 u_2 =wind speed 2 m above water surface (m/s)

University of Babylon

College of Engineering



The overall energy budget for a lake can be written as

$$\mathbf{Q}_{\mathbf{N}} = \mathbf{Q}_{\mathbf{e}} + \mathbf{Q}_{\mathbf{h}} - \mathbf{Q}_{\mathbf{v}} + \mathbf{Q}_{\mathbf{\theta}}$$

 Q_N : net radiation absorbed by water body [cal/cm2-day]

(solar radiation -reflection -radiation from lake)

Qe: evaporation energy

Q_h: sensible heat transfer (conduction and convection to the atmosphere)

 Q_v : advected energy of inflow and outflow

 $Q_{\boldsymbol{\theta}} \text{:}$ change in stored energy in the water body the overall energy budget

Sensible heat transfer difficult to measure

 $Q_h \approx R \; x \; Q_e$

$$\mathbf{R} = \gamma \; \frac{\mathbf{T}_{\mathbf{s}-}\mathbf{T}_{\mathbf{a}}}{\mathbf{e}_{\mathbf{s}-}\mathbf{e}_{\mathbf{a}}}$$

T_a: air temperature [°C]

 T_s : water surface temperature [°C]

e_a: vapor pressure of the air [mb]

e_s: saturation vapor pressure at water surface temp. [mb]

University of Babylon

College of Engineering

 γ : psychrometric constant = 0.66 (P/1000), P: atmospheric pressure in mb

Daily evaporation depth: $\mathbf{E} = \gamma \frac{\mathbf{Q}_{\mathbf{e}}}{\rho \mathbf{L}_{\mathbf{e}}}$ cm/day

Energy balance
$$Q_N = Q_e + Q_h - Q_v + Q_\theta$$

 $Q_h = R * Q_e$
 $Q_N = Q_e * (1+R) - Q_v + Q_\theta$
 $Q_N = E\rho L_e (1+R) - Q_v + Q_\theta$
 $E = \frac{Q_N + Q_v - Q_\theta}{\rho L_e (1+R)}$ cm/day

with

Q in $[cal/cm^2-day]$

 L_e in [cal/g]: the latent heat of vaporization ρ in [g/cm³]: mass density of evaporated water •Most accurate method

3. Combined method (Penman, 1948)

Combined 'mass transfer' and 'energy budget':

$$\mathbf{E} \ \boldsymbol{\rho} \mathbf{L}_{\mathbf{e}} = \frac{\Delta}{\Delta + \gamma} \ \mathbf{Q}_{\mathbf{N}} + \frac{\gamma}{\Delta + \gamma} \ \mathbf{E}_{\mathbf{a}}$$

 $L_e = 597.3 - 0.57 (T-0 C^0)$

 Δ : slope of e_s vs t curve or

$$\Delta = \frac{de_{\rm s}}{d_{\rm T}} = \frac{2.7489 * 10^8 * 4278.6}{(T+242.79)^2} e^{\left(\frac{-4278.6}{T+242.79}\right)}$$

 Q_N = is net absorbed radiation

E_a: drying power from equation

$$\gamma = 0.66 * \left(\frac{P}{1000}\right)$$
$$\mathbf{E_a} = \rho \mathbf{L_e} (\mathbf{a} + \mathbf{bu}) (\mathbf{e_{sa}} - \mathbf{e_a}) \qquad \text{cal/cm}^2 - \text{day}$$

where

a,b : empirical constants

University of Babylon

College of Engineering

e_{sa}: saturation vapor pressure at air temp.

e_a: actual vapor pressure

Example :

Assume Meyer's formula applies to a lake:

 $E = 0.0106 (1+0.1 u)(e_s-e_a) [cm/day]$

u in mi/h, e in mb

Given:

 $T_a = 32.2^{\circ}C$

u = 32 km/h = 20 mi/h

RH = 30%

QN = 400 cal/cm2-day

Estimate daily evaporation using Penman's formula.

Solution

$$\Delta = \frac{2.7489 \times 10^8 \times 4278.6}{(T+242.79)^2} \exp\left\{-\frac{4278.6}{(T+242.79)}\right\}$$

with T = 32.2° c, $\Delta = 2.72 \text{ mb/}^{\circ}$ c

Actual and saturation vapor pressure:

 $e_{sa} (T = 32.2^{\circ}c) = 48.1 \text{mb}$

 $e_a = RH \ge e_{sa} = 0.3 \ge 48.1 = 14.4 mb$

Latent heat of evaporation at air temperature:

$$Le = 597.3 - 0.57 \text{ x } 32.2 = 579 \text{ cal/g}$$

$$E_a = \rho L_e 0.0106 (1+0.1u) (e_{sa}-e_a)$$

$$= 1 \times 579 \times 0.0106 (1+0.1 \times 20) (48.1-14.4)$$

 $= 1590 \text{ cal/cm}^2 \text{-day}$

Penman's equation:

$$\mathbf{E} \ \boldsymbol{\rho} \mathbf{L}_{\mathbf{e}} = \frac{\Delta}{\Delta + \gamma} \mathbf{Q}_{\mathbf{N}} + \frac{\gamma}{\Delta + \gamma} \mathbf{E}_{\mathbf{a}}$$
$$= \frac{2.72}{2.72 + 0.66} \ 400 + \frac{0.66}{2.72 + 0.66} \ 1590$$

University of Babylon

College of Engineering

Hydrology Lecture 7 Anees Kadhum AL – Saadi M.Sc. Water Resources

 $= 632 \text{ cal/cm}^2\text{-day}$

or

 $E = \frac{632}{1 x 579} = 1.1 \text{ cm/day}$

4: Water budget method

Applicable to lake evaporation

 Δ storage = input – output

 $\Delta S = (I+P)-(O+E+GW)$

Or

 $E = -\Delta S + I + P - O - GW$

I : inflow [cm]

P: precipitation [cm]

O: outflow [cm]

E : Evaporation [cm]

GW: Groundwater seepage [cm]