

Glycosides

Introduction

■ Definition:

- Glycosides are non-reducing organic compounds that on hydrolysis with acids, alkalis or enzymes yield:
 - A sugar part (or glycone, formed of one or more sugar units).
 - A non-sugar part (or aglycone, also called genin).

Classification:

■ Atom from the aglycone involved in the glycosidic linkage:

- | | |
|----------------------|--------------|
| ■ Aglycone- O- Sugar | O-glycosides |
| ■ Aglycone- C- Sugar | C-glycosides |
| ■ Aglycone- S- Sugar | S-glycosides |
| ■ Aglycone- N- Sugar | N-glycosides |

■ Number of sugars:

- | | | |
|----------------|-----------|---------------|
| ■ One sugar | monosides | e.g. Salicin. |
| ■ Two sugar | Biosides | e.g. Diosmin. |
| ■ Three sugars | Triosides | e.g. Digoxin. |

■ Nature of the glycoside:

- | | |
|--|-----------------|
| ■ Primary glycosides: Originally present in the plant | e.g. Purpurea A |
| ■ Secondary glycosides: Resulted from removal of one sugar from the primary glycosides | e.g. Digitoxin |

- **Type of the glycosidic linkage:**
 - α - glycosides
 - β - glycosides
- **Botanical source:**
 - Digitalis glycosides
 - Senna glycosides.
- **Therapeutic use:**
 - Analgesic glycosides.
 - Purgative glycosides.
 - Cardiac glycosides.
- **Chemical nature of the aglycone:**
 - Flavone glycosides.
 - Steroidal glycosides.
 - Aldehydic glycosides.

Physical Characters:

- Solids either amorphous or crystalline.
- Non volatile.
- Usually bitter in taste.
- Soluble in water and polar organic solvents.
- Reduce Fehling's solutions only after hydrolysis.

Stability of Glycosides:

1- Effect of acid hydrolysis:

- Acids split sugars from the aglycones.
- The acetal linkage is more readily cleaved than the linkage between the individual sugars of the sugar chain.
- C-glycosides are resistant to acid hydrolysis.

2- Effect of alkaline hydrolysis:

A- Strong alkalis:

- Hydrolysis of ester groups.
- Opening of lactone rings e.g. Cardiac glycosides.

B- Mild alkalis:

- Hydrolysis of ester groups e.g. Lanatoside A to Purpurea A
- Opening of lactone rings e.g. Cardiac glycosides.

3- Enzymatic hydrolysis:

- Split the sugars stepwise starting from the terminal sugars.
- All plants producing glycosides have enzyme that can hydrolyze these glycosides.
- Enzymes are specific for the type of glycosidic linkages:
 - Emulsin can hydrolyze β - glycosides
 - Invertase can hydrolyze α - glycosides
 - Myrosin can hydrolyze s-glycosides.

Extraction and Isolation

- Because of the wide range of physical and chemical properties of glycosides and other constituents associated with them, no common general method for their isolation is recommended.
- Water, methanol, water-ethanol and ethanol are the most common solvents for extraction of glycosides.
- Precautions before extraction
 - Deactivation of enzymes:
 - **Drying** for 15-30 min at **100 °C** followed by slow drying at a low temperature.
 - Dipping the fresh material into **boiling water** or **boiling alcohol** for 10-20 min.
 - **Boiling** the fresh plant material with **acetone**.
 - Carrying out the extraction at very low temp.
 - **Freeze-drying** of the plant material **before extraction** (lyophilization).
 - Carrying the extraction in the presence of $(\text{NH}_4)_2\text{SO}_4$.

■ Maintenance of neutral conditions:

- Neutral pH should be assured before and during extraction because:
- Acidity may result in hydrolysis. This is overcome by addition of CaCO_3 .
- Mild alkalinity may sometimes produce racemization.

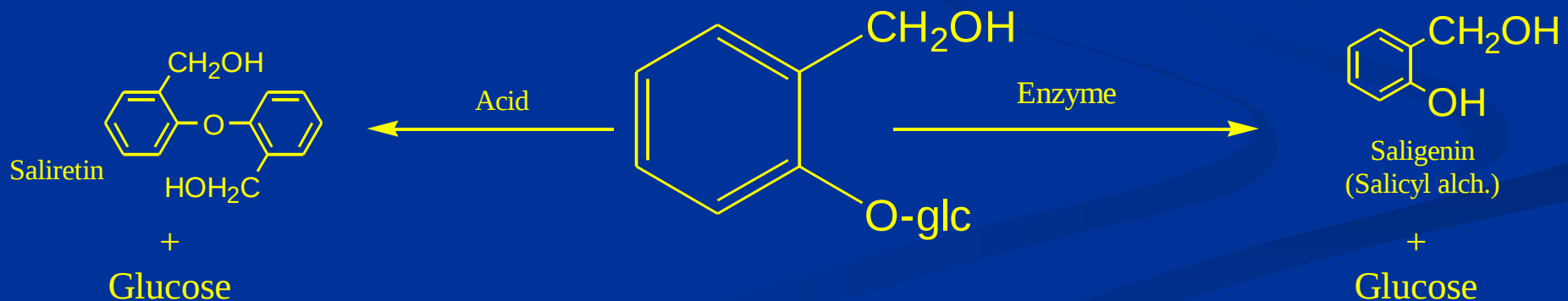
■ Defatting of fat-rich organs (e.g. seeds) before extraction:

- High amounts of lipoids hinder glycoside extraction. Defatting is usually carried with petroleum ether

Alcoholic and Phenolic Glycosides

1- Salicin

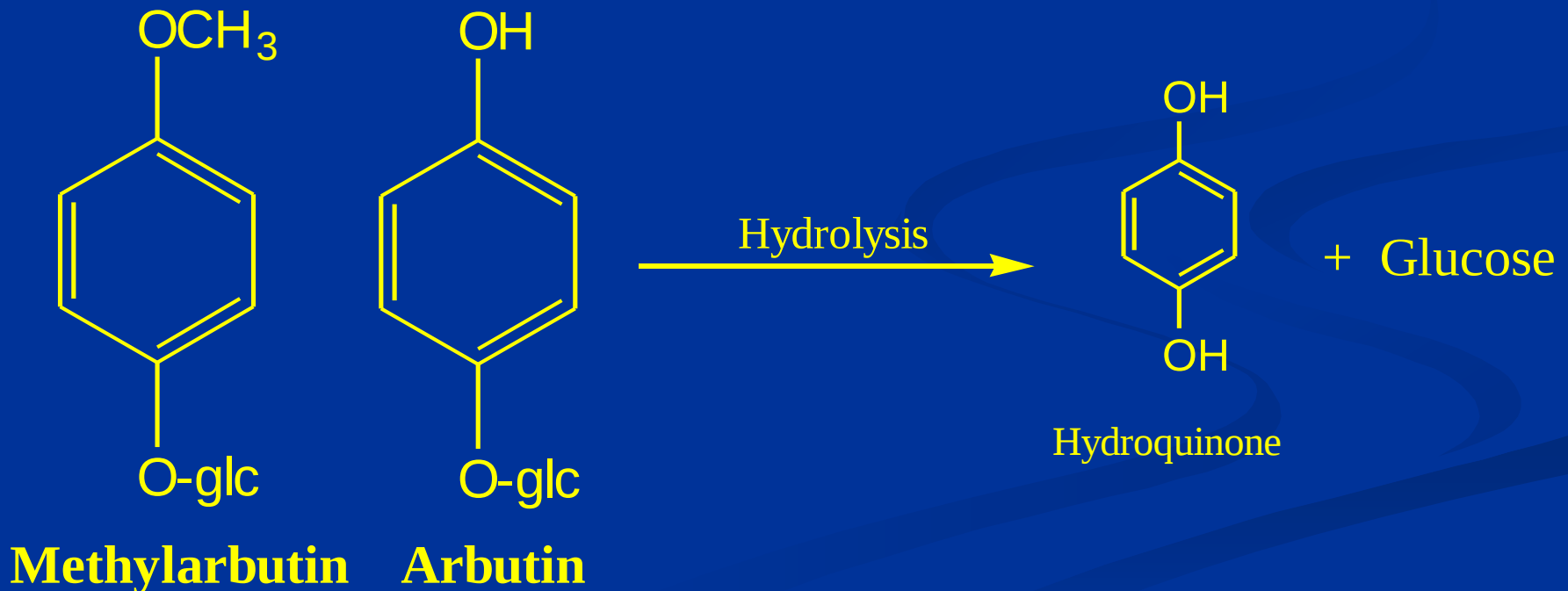
- **Source:** Salix species (Willow bark).
- **Nature:** Primary alcoholic and Phenolic glycoside (monoside).
- **Uses:** Analgesic- Antipyretic- Anti-inflammatory.

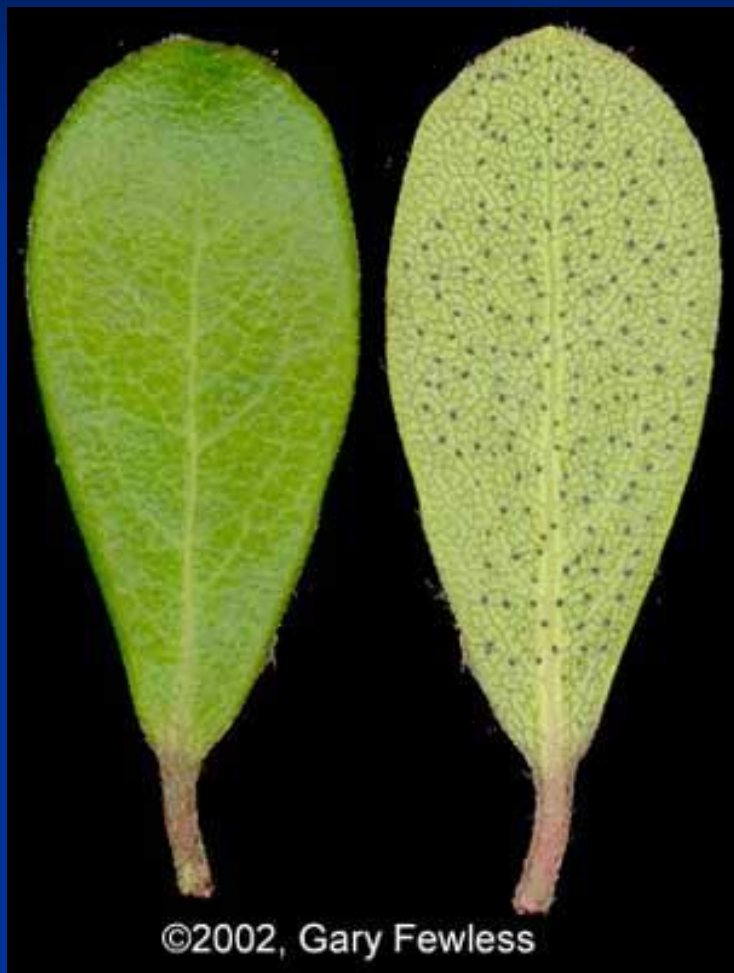




2- Arbutin & Methyl Arbutin

- **Source:** Uva Ursi (Bearberry leaves).
- **Nature:** Primary Phenolic glycoside (monoside).
- **Uses:** Diuretic- Bactericidal.



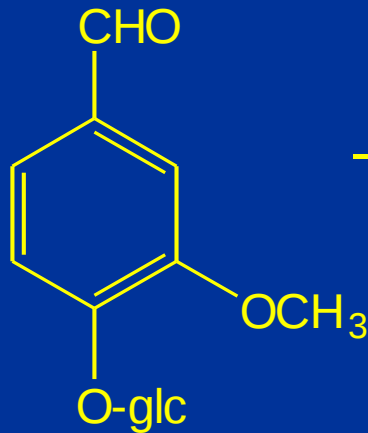


Aldehydic Glycosides

1- Glucovanillin

- **Source:** Vanilla pods.
- **Uses:** Flavouring agent- Spray reagent.

Glucovanillin

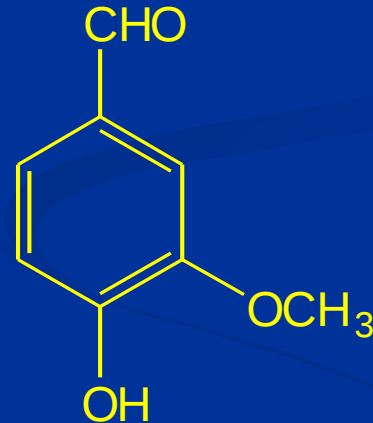


Green vanilla pods
Bitter in taste
Odourless

Enzymatic Hydrolysis



Vanillin

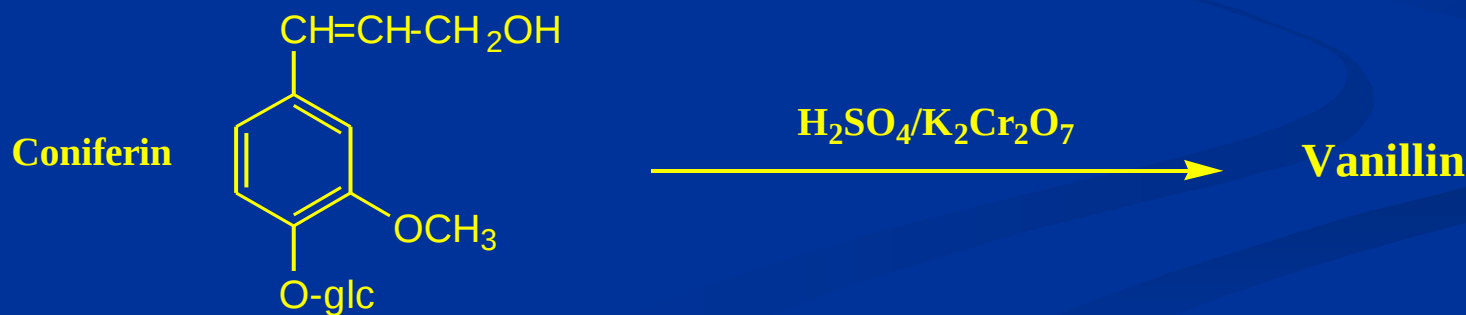
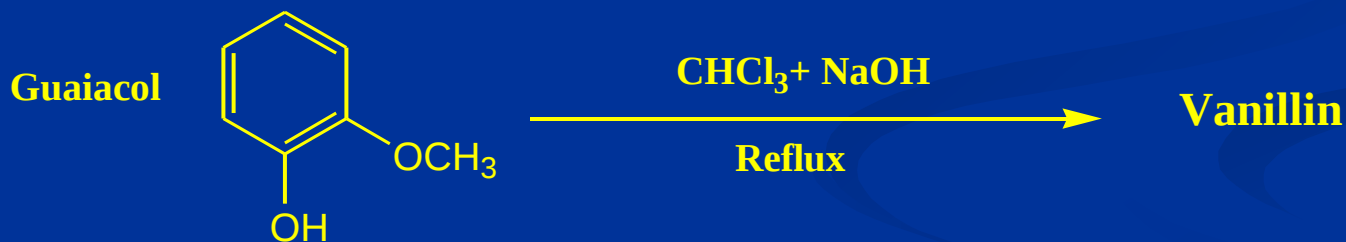
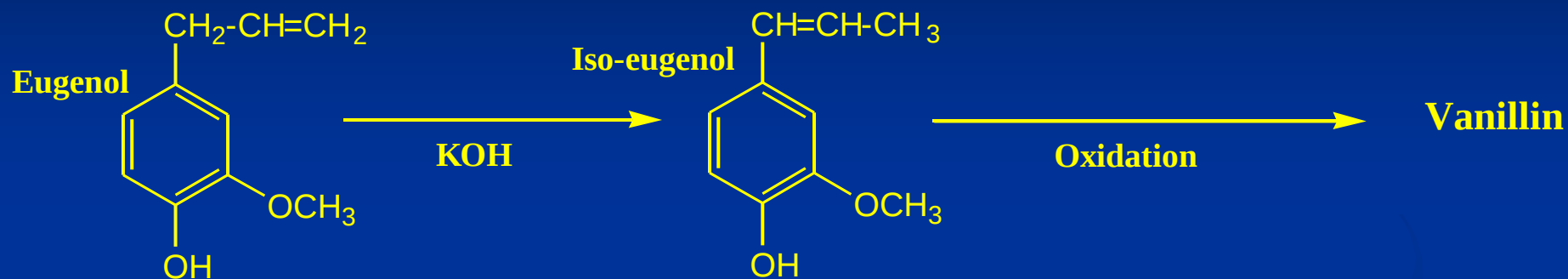


Brown vanilla pods
Sweet in taste
Vanilla odour

+ Glucose

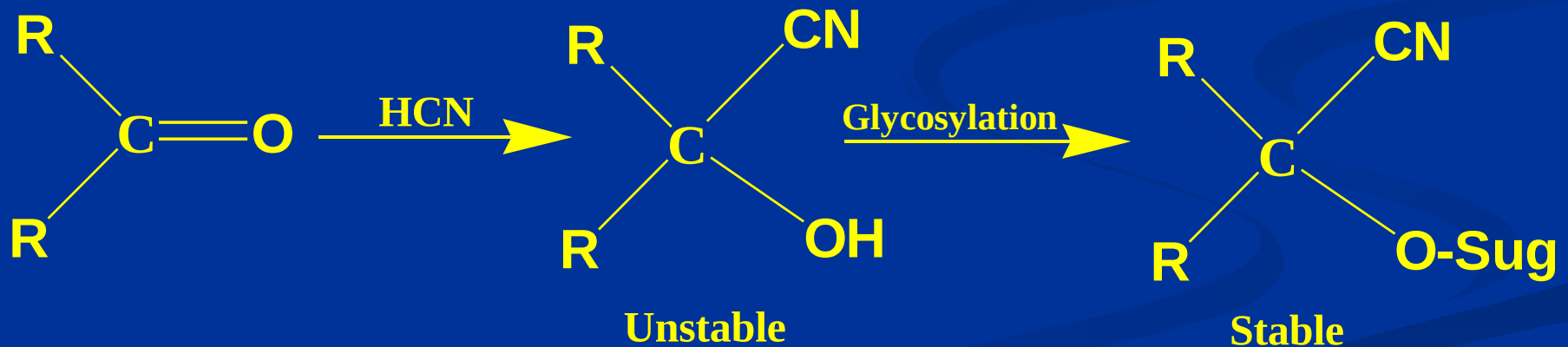


Commercial Preparation of Vanillin



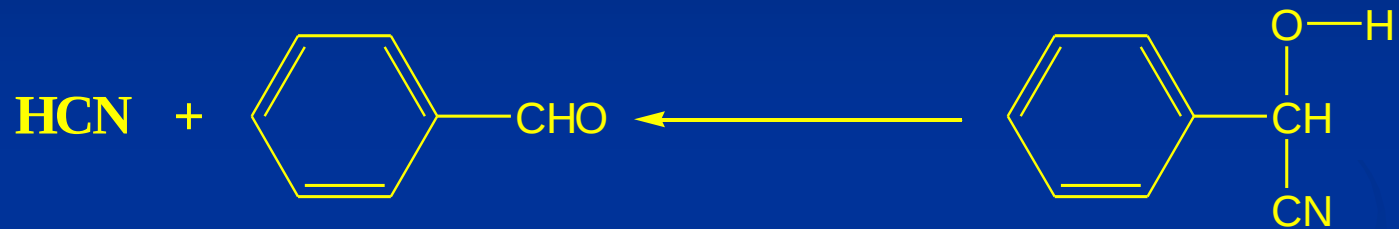
Cyanogenic Glycosides

- **Cyanogenic glycosides** (Cyanogentic or Cyanophore Glycosides) are **O-glycosides** yielding **HCN** gas on **hydrolysis**.
- They are condensation products of HCN to a carbonyl compounds (Cyanohydrin).

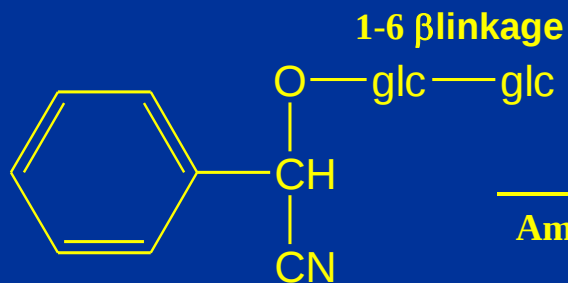


1- Amygdalin

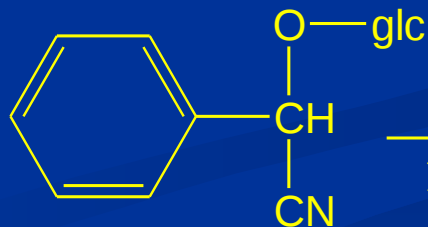
- **Source:** Bitter Almond.
- **Structures:** It is a Bioside of mandilonitril.



Mandilonitril



Amygdalin



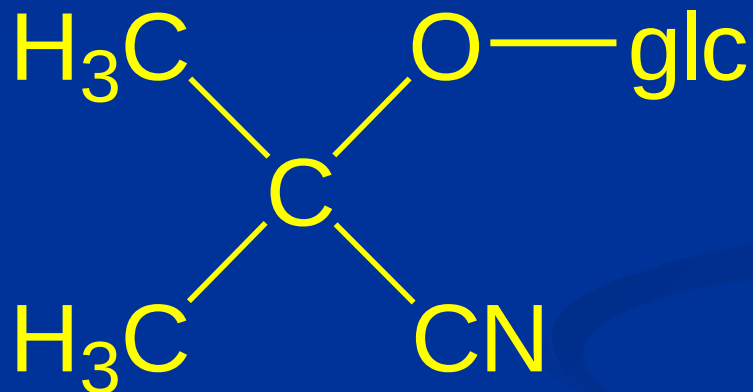
Prunasin

Prunase



2- Linamarin

- **Source:** Linseed.
- **Structures:** It is the glycosidic derivative of the cyanohydrin of acetone.



- **Uses:**
 - Linamarin has a molluscecidal activity.
 - Amygdalin is used for the preparation of Benzaldehyde.
 - Cyanogenic glycosides have role in cancer treatment.



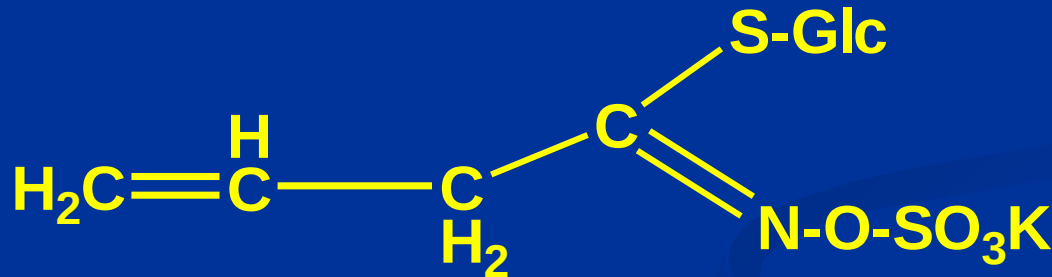
Test for Cyanogenic Glycosides:

- Reduce plant material to small pieces and moisten with water.
- Incubate at temp. less than 45 °C for 30 60 mins with the neck of the flask stoppered and have suspended sodium picrate paper.
- The paper will turn brick red due to the release of HCN gas.

Thioglycoside

Glucosinolates- Sulphur Glycosides

- They are S-glycosides widely distributed in family Cruciferae.
- **Sinigrin:** In seeds of *Brassica nigra* (black mustard).
- **Sinalbin:** In Seeds of *Brassica alba* (white mustard).



Sinigrin

- **Uses:** Rubefacients, Counter irritants and condiment.



Garlic

- It consists of the bulb of *Allium sativum* Fam. Liliaceae.
- The intact cells of garlic contain an odorless, sulfur-containing amino acid derivative (+)-*S*-allyl-*L*-cysteine sulfoxide, commonly known as alliin.
- Alliin is hydrolyzed by the effect of alliinase enzyme present in different cells after crushing into allicin (diallyl thiosulfinate).
- Allicin is responsible for the characteristic odor and flavor of garlic.
- Allicin is a potent antibacterial, antihyperlipidemic, and it inhibits platelet aggregation and enhances the blood fibrinolytic activity.

