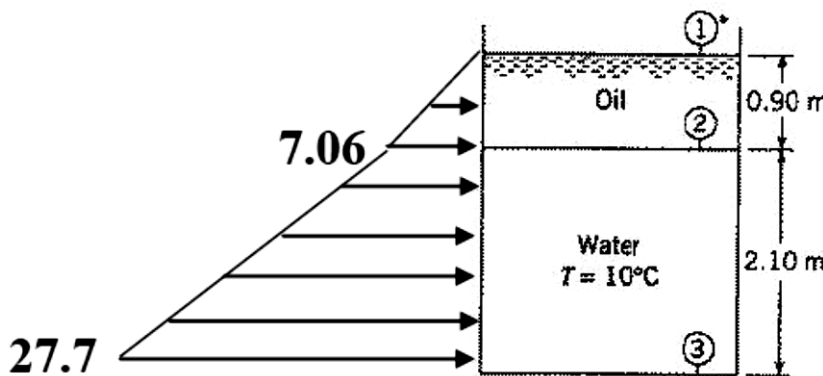


Problems on pressure variation and scales

Example 1:

Oil with a specific gravity of 0.80 forms a layer 0.90 m deep in an open tank that is otherwise filled with water. The total depth of water and oil is 3 m. What is the gage pressure at the bottom of the tank?



$$p + \gamma Z = \text{constant}$$

$$p_1 + \gamma Z_1 = p_2 + \gamma Z_2$$

$$p_2 = p_1 + \gamma(Z_1 - Z_2)$$

$$p_1 = p_{\text{atm}} = 0$$

$$p_2 = \gamma_{\text{oil}} \Delta Z = 0.8 \times 9810 \times 0.9 = 7.06 \text{ kPa}$$

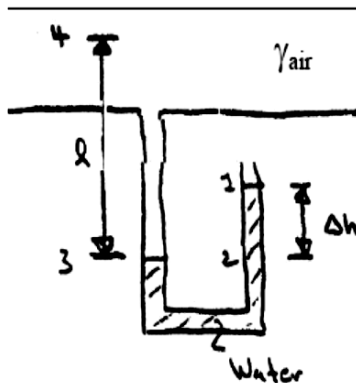
$$p_3 = p_2 + \gamma_{\text{water}}(Z_2 - Z_3)$$

$$= 7060 + 9810 \times 2.1$$

$$= 27.7 \text{ kPa}$$

Example:

Air at 20°C is in pipe with a water manometer. For given conditions compute gage pressure in pipe.



$$l = 140 \text{ cm}$$

$$\Delta h = 70 \text{ cm}$$

$$p_4 = ? \text{ gage (i.e., } p_1 = 0)$$

$$p_1 + \gamma \Delta h = p_3$$

$$p_3 - \gamma_{\text{air}} l = p_4$$

step-by-step method

Pressure same at 2&3 since same elevation & Pascal's law: in closed system pressure change produced at one part transmitted throughout entire system

$$p_1 + \gamma \Delta h - \gamma_{\text{air}} l = p_4 \quad \text{complete circuit method}$$

$$\gamma \Delta h - \gamma_{\text{air}} l = p_4 \quad \text{gage}$$

$$\gamma_{\text{water}}(20^\circ\text{C}) = 9790 \text{ N/m}^3 \Rightarrow p_3 = \gamma \Delta h = 6853 \text{ Pa [N/m}^2]$$

$$\gamma_{\text{air}} = \rho g$$

$$\rho = \frac{p}{RT} = \frac{(p_3 + p_{\text{atm}})}{R(^{\circ}\text{C} + 273)} = \frac{6853 + 101300}{287(20 + 273)} = 1.286 \text{ kg/m}^3$$

\swarrow p_{abs}
 \swarrow $^{\circ}\text{K}$

$$\gamma_{\text{air}} = 1.286 \times 9.81 \text{ m/s}^2 = 12.62 \text{ N/m}^3$$

note $\gamma_{\text{air}} \ll \gamma_{\text{water}}$

$$p_4 = p_3 - \gamma_{\text{air}} l = 6853 - \underbrace{12.62 \times 1.4}_{17.668} = 6835 \text{ Pa}$$

if neglect effect of air column

$$p_4 = 6853 \text{ Pa}$$

2.46 In Fig. P2.46 both ends of the manometer are open to the atmosphere. Estimate the specific gravity of fluid X.

Solution: The pressure at the bottom of the manometer must be the same regardless of which leg we approach through, left or right:

$$p_{\text{atm}} + (8720)(0.1) + (9790)(0.07) + \gamma_X(0.04) \quad (\text{left leg})$$

$$= p_{\text{atm}} + (8720)(0.09) + (9790)(0.05) + \gamma_X(0.06) \quad (\text{right leg})$$

$$\text{or: } \gamma_X = 14150 \text{ N/m}^3, \quad \text{SG}_X = \frac{14150}{9790} \approx 1.45 \quad \text{Ans.}$$

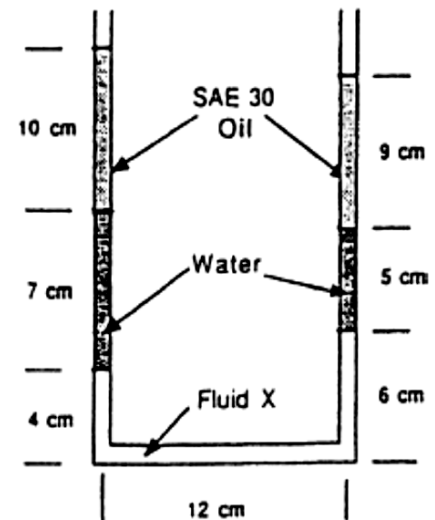


Fig. P2.46

2.41 The system in Fig. P2.41 is at 20°C. Determine the pressure at point A in pounds per square foot.

Solution: Take the specific weights of water and mercury from Table 2.1. Write the hydrostatic formula from point A to the water surface:

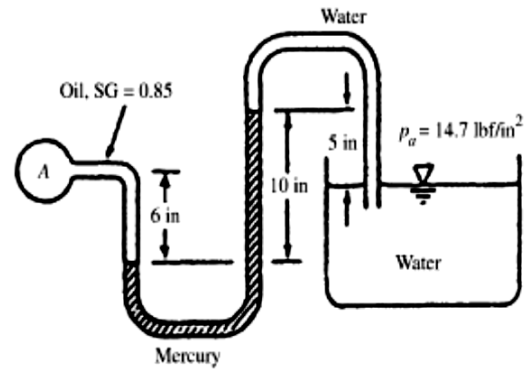


Fig. P2.41

$$p_A + (0.85)(62.4 \text{ lbf/ft}^3) \left(\frac{6}{12} \text{ ft} \right) - (846) \left(\frac{10}{12} \right) + (62.4) \left(\frac{5}{12} \right) = p_{\text{atm}} = (14.7)(144) \frac{\text{lbf}}{\text{ft}^2}$$

$$\text{Solve for } p_A = 2770 \text{ lbf/ft}^2 \quad \text{Ans.}$$

2.34 To show the effect of manometer dimensions, consider Fig. P2.34. The containers (a) and (b) are cylindrical and are such that $p_a = p_b$ as shown. Suppose the oil-water interface on the right moves up a distance $\Delta h < h$. Derive a formula for the difference $p_a - p_b$ when (a) $d \ll D$; and (b) $d = 0.15D$. What is the % difference?

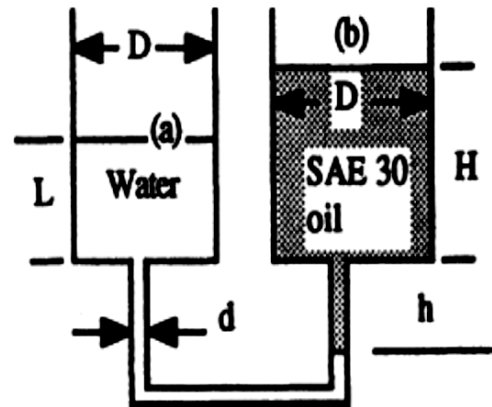


Fig. P2.34

Solution: Take $\gamma = 9790 \text{ N/m}^3$ for water and 8720 N/m^3 for SAE 30 oil. Let “H” be the height of the oil in reservoir (b). For the condition shown, $p_a = p_b$, therefore

$$\gamma_{\text{water}}(L + h) = \gamma_{\text{oil}}(H + h), \quad \text{or: } H = (\gamma_{\text{water}}/\gamma_{\text{oil}})(L + h) - h \quad (1)$$

Case (a), $d \ll D$: When the meniscus rises Δh , there will be no significant change in reservoir levels. Therefore we can write a simple hydrostatic relation from (a) to (b):

$$p_a + \gamma_{\text{water}}(L + h - \Delta h) - \gamma_{\text{oil}}(H + h - \Delta h) = p_b,$$

$$\text{or: } p_a - p_b = \Delta h(\gamma_{\text{water}} - \gamma_{\text{oil}}) \quad \text{Ans. (a)}$$

where we have used Eq. (1) above to eliminate H and L. Putting in numbers to compare later with part (b), we have $\Delta p = \Delta h(9790 - 8720) = 1070 \Delta h$, with Δh in meters.

Case (b), $d = 0.15D$. Here we must account for reservoir volume changes. For a rise $\Delta h < h$, a volume $(\pi/4)d^2\Delta h$ of water leaves reservoir (a), decreasing "L" by $\Delta h(d/D)^2$, and an identical volume of oil enters reservoir (b), increasing "H" by the same amount $\Delta h(d/D)^2$. The hydrostatic relation between (a) and (b) becomes, for this case,

$$p_a + \gamma_{\text{water}}[L - \Delta h(d/D)^2 + h - \Delta h] - \gamma_{\text{oil}}[H + \Delta h(d/D)^2 + h - \Delta h] = p_b,$$

$$\text{or: } p_a - p_b = \Delta h[\gamma_{\text{water}}(1 + d^2/D^2) - \gamma_{\text{oil}}(1 - d^2/D^2)] \quad \text{Ans. (b)}$$

where again we have used Eq. (1) to eliminate H and L. If d is not small, this is a *considerable* difference, with surprisingly large error. For the case $d = 0.15 D$, with water and oil, we obtain $\Delta p = \Delta h[1.0225(9790) - 0.9775(8720)] \approx 1486 \Delta h$ or 39% more than (a).

2.13 In Fig. P2.13 the 20°C water and gasoline are open to the atmosphere and are at the same elevation. What is the height h in the third liquid?

Solution: Take water = 9790 N/m³ and gasoline = 6670 N/m³. The bottom pressure must be the same whether we move down through the water or through the gasoline into the third fluid:

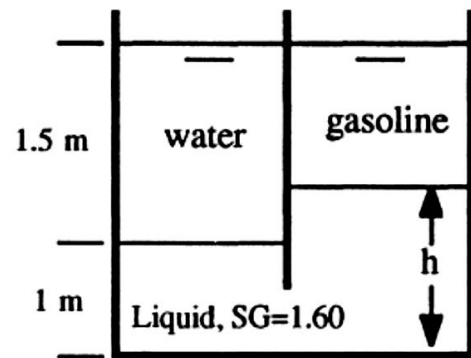


Fig. P2.13

$$p_{\text{bottom}} = (9790 \text{ N/m}^3)(1.5 \text{ m}) + 1.60(9790)(1.0) = 1.60(9790)h + 6670(2.5 - h)$$

$$\text{Solve for } h = 1.52 \text{ m} \quad \text{Ans.}$$

Exercises:

- 1 Express standard atmospheric pressure as a head, $h = p/\rho g$, in (a) feet of ethylene glycol; (b) inches of mercury; (c) meters of water; and (d) mm of methanol.

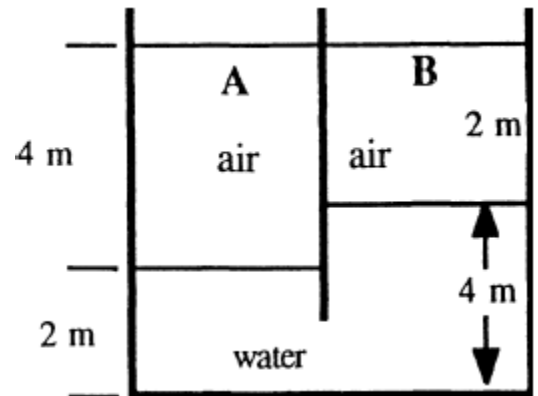
Ans: a- 30.3 ft; b- 30.0 in; c- 10.35 m; d- 13.13m.

2. A closed tank contains 1.5 m of SAE 30 oil, 1 m of water, 20 cm of mercury, and an air space on top, all at 20°C. If $p_{\text{bottom}} = 60 \text{ kPa}$, what is the pressure in the air space?

Ans: 10500 Pa.

3. The closed tank in Fig. P2.14 is at 20°C. If the pressure at A is 95 kPa absolute, determine p at B (absolute). What percent error do you make by neglecting the specific weight of the air?

Ans: 75420 Pa.



4. In Fig. P2.21 all fluids are at 20°C. Gage A reads 350 kPa absolute. Determine (a) the height h in cm; and (b) the reading of gage B in kPa absolute.

Ans: a- $h = 6.49 \text{ m}$

b- $P_B = 251 \text{ kPa}$.

